# A Selective Synthesis of 2-Alkylamino-5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones 

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#### Abstract

Alkylamino-5, 6, 7, 8-tetrahydrobenzothieno [2, 3-d] pyrimidin-4(3H)-ones 7 were synthesized by a new selective synthetic method, which includes aza-Wittig reaction of iminophosphorane $\mathbf{4}$ with aromatic isocynate to give carbodiimide 5 and subsequent reaction of 5 with various aliphatic primary amine in the presence of $\mathrm{EtO} \mathrm{Na}^{+}$.


Keywords: Benzothieno[2, 3-d]pyrimidin-4(3H)-ones, aza-Wittig reaction, iminophosphorane, carbodiimide, synthesis.

Thienopyrimidines are of great importances because of their remarkable biological properties. For example, some 2 -substituted thienopyrimidinones show significant antifungal and antibacterial activities ${ }^{1-3}$, whereas others exhibited good anticonvulsant and angiotensin II or $\mathrm{H}_{1}$ receptor antagonistic activities ${ }^{4-6}$. There are many known methods for the synthesis of thienopyrimidinones ${ }^{7-9}$, however, 2-amino substituted thienopyrimidinones were not easily accessible by currently existing routes. Recently we have been interested in the synthesis of imidazolinones, quinazolinones and thienopyrimidinones via aza-Wittig reaction, with the aim of evaluating their fungicidal activities ${ }^{10-13}$. We have also reported a selective synthesis of 2-alkylaminoimidazolones and 2-alkylaminoquinazolinones by reaction of aliphatic primary amine with functionalized carbodiimides ${ }^{13-15}$. This method was further utilized in the synthesis of 2 -alkylamino substituted 5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones 7 from easily accessible iminophosphorane 4.

The tetrahydrobenzo[b]thiophene 2, easily obtained by Gewald method from cyclohexanone 1, ethyl cyanoacetate and sulfur ${ }^{16}$, was converted to iminophosphorane 4 via reaction with triphenylphosphine, hexachloroethane and triethylamine ${ }^{17}$. Iminophosphorane 4 could also be prepared by Staudinger reaction of triphenylphosphine with azide 3 which was obtained by first diazotization of tetrahydrobenzo[b]thiophene 2 with $\mathrm{NaNO}_{2} / \mathrm{HCl}$ and subsequent azidation with $\mathrm{NaN}_{3}$ (Scheme 1).

[^0]Scheme 1


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(a) $\mathrm{EtOOCCH} \mathrm{CN}_{2} \mathrm{CN}$, S, morpholine, $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$, r.t., 12 h, $76 \%$; (b) $\mathrm{Ph}_{3} \mathrm{P}, \mathrm{C}_{2} \mathrm{Cl}_{6}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., 5 h, $78 \%$. (c) $\mathrm{NaNO}_{2}, \mathrm{HCl}$, actone- $\mathrm{H}_{2} \mathrm{O}$, then $\mathrm{NaN}_{3}, \mathrm{NaAc}, 25 \%$; (d) $\mathrm{Ph}_{3} \mathrm{P}^{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., 6 h, $85 \%$.

Scheme 2

(e) $\mathrm{ArNCO}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0-5^{\circ} \mathrm{C}$, 12 h ; (f) $\mathrm{RNH}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., 10 min ; (g) EtONa, EtOH, r.t., 6-12 h, 51-83\%.

Table 1 Preparation of 2-alkylaminothieno[2, 3-d]pyrimidin-4(3H)-ones 7

|  | Ar | R | Condition | Yield (\%) | $\mathrm{Mp}\left({ }^{\circ} \mathrm{C}\right)$ | Elementary analysis (\%, Calcd.) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  | C | H | N |
| 7a | Ph | $n-\mathrm{C}_{3} \mathrm{H}_{7}$ | r.t./6 hr | 55 | 132-134 | 67.38(67.23) | 6.20(6.24) | 12.59(12.38) |
| 7b | Ph | $i-\mathrm{C}_{3} \mathrm{H}_{7}$ | r.t. $/ 8 \mathrm{hr}$ | 67 | 155-157 | $67.25(67.23)$ | 6.11(6.24) | 12.24(12.38) |
| 7c | Ph | cyclohexyl | r.t. $/ 8 \mathrm{hr}$ | 72 | 181-183 | 69.54(69.62) | 6.47(6.64) | 11.12(11.07) |
| 7d | Ph | $t-\mathrm{C}_{4} \mathrm{H}_{9}$ | r.t./12 hr | 83 | 182-184 | 67.81(67.96) | 6.62(6.56) | 11.93(11.89) |
| 7 C | 4-Cl- $\mathrm{C}_{6} \mathrm{H}_{4}$ | $n-\mathrm{C}_{3} \mathrm{H}_{7}$ | r.t./6 hr | 51 | 185-187 | 61.20(61.03) | 5.35(5.39) | 11.31(11.24) |
| 7f | 4-Cl-C6 $\mathrm{H}_{4}$ | $n-\mathrm{C}_{4} \mathrm{H}_{9}$ | r.t./6 hr | 54 | 132-134 | 61.78(61.92) | 5.74(5.72) | 10.89(10.83) |
| 7 g | 4-Cl-C6 $\mathrm{H}_{4}$ | $i-\mathrm{C}_{3} \mathrm{H}_{7}$ | r.t./10 hr | 60 | 190-191 | 61.12(61.03) | 5.47(5.39) | 11.12(11.24) |
| 7h | 4-Cl-C6 $\mathrm{H}_{4}$ | cyclohexyl | r.t. $/ 10 \mathrm{hr}$ | 68 | 186-187 | 63.75(63.83) | $5.90(5.84)$ | 11.34(10.15) |
| 71 | 4-Cl-C6 $\mathrm{H}_{4}$ | $t-\mathrm{C}_{4} \mathrm{H}_{9}$ | r.t./12 hr | 81 | 242-244 | 61.84(61.92) | 5.65(5.72) | 10.93(10.83) |
| 7 j | 4-Me-C6 $\mathrm{H}_{4}$ | $i-\mathrm{C}_{3} \mathrm{H}_{7}$ | r.t. $/ 8 \mathrm{hr}$ | 75 | 139-141 | 67.94(67.96) | 6.68(6.56) | 11.96(11.89) |

Iminophosphorane 4 reacted with aromatic isocyanates to give carbodiimides 5, which were allowed to react with aliphatic primary amines in the presence of EtO- Na ${ }^{+}$to provide only 2-alkylamino-5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones 7, one of the possible regioisomers (Scheme 2). Different from the early result in similar cases ${ }^{13-15}$, we obtained only 7 from the reaction mixture after recrystallization; the other isomer was not found by GC-MS detection or ${ }^{1} \mathrm{H}$ NMR analysis of the reaction
mixture. The structure of 7 is deduced from their ${ }^{1} \mathrm{H}$ NMR data. For example, the ${ }^{1} \mathrm{H}$ NMR spectrum in $7 \mathbf{7}$ shows the signals of NH at 4.04 ppm as triple absorption and $\mathrm{NCH}_{2}$ at $3.35-3.29 \mathrm{ppm}$ as multiple absorption, which strongly suggest the existence of $\mathrm{NHCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ group in 7a. Moreover, when the sample was treated with deuterated water, the ${ }^{1} \mathrm{H}$ NMR spectrum in 7 a shows the absorption of $\mathrm{NCH}_{2}$ at 3.32 ppm as triplets with the disappearance of signals of NH absorption. Whenever the primary amine used is small $(\mathrm{R}=n-\mathrm{Pr})$ or bulky $(\mathrm{R}=t-\mathrm{Bu})$, the cyclization was achieved all in moderate to good yields with the similar selectivity. The results are listed in Table 1. The solitary formation of 7 can be rationalized in terms of an initial nucleophilic addition to give the guanidine intermediate 6 , which cyclized to give 7 across the arylamino group rather than the alkylamino one. This may probably due to the preferential generation of $-\mathrm{N}^{-} \mathrm{Ar}$ from more acidic-NHAr under the catalysis of $\mathrm{EtO}^{-} \mathrm{Na}^{+}$.

Table 2 IR, MS and ${ }^{1} \mathrm{H}$ NMR of 7

|  | $\begin{gathered} \text { IR (KBr, } \\ \left.\mathrm{cm}^{-1}\right) \\ \hline \end{gathered}$ | MS ( $m / \mathrm{z}, \%$ ) | ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}, \delta \mathrm{ppm}\right)$ |
| :---: | :---: | :---: | :---: |
| 7a | $\begin{aligned} & \hline 3431,1677, \\ & 1556,1240 \end{aligned}$ | $\begin{aligned} & 339\left(\mathrm{M}^{+}, 100\right), 311(33), \\ & 297(77), 269(62), 220 \\ & (32), 77(52) \end{aligned}$ | 7.60-7.26 (m, 5H, Ar-H), 4.04 (t, 1H, J=4.8 Hz, NH), 3.35-3.29 (m, $2 \mathrm{H}, \mathrm{NCH}_{2}$ ), 2.90-2.67 (m, $4 \mathrm{H}, 2 \mathrm{CH}_{2}$ ), 1.88-1.76 (m, 4H, $2 \mathrm{CH}_{2}$ ), 1.52-0.81 ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) |
| 7b | $\begin{aligned} & 3328,1672, \\ & 1544,1172 \end{aligned}$ | $\begin{aligned} & 339\left(\mathrm{M}^{+}, 100\right), 311(38), \\ & 297(34), 269(65), 221 \\ & (33), 77(56) \end{aligned}$ | 7.61-7.25 (m, 5H, Ar-H), 4.25-4.15 (m, 1H, NCH), $3.81(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}, \mathrm{NH}), 2.90-2.67\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right)$, 1.86-1.76 (m, 4H, 2CH2), $1.10\left(\mathrm{~d}, 6 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}, 2 \mathrm{CH}_{3}\right)$ |
| 7c | $\begin{aligned} & 3427,1683, \\ & 1545,1236 \end{aligned}$ | $\begin{aligned} & 379\left(\mathrm{M}^{+}, 24\right), 297(58), \\ & 269(71), 179(18), 77 \\ & (100) \end{aligned}$ | 7.65-7.26 (m, 5H, Ar-H), 4.00-3.88 (m, 2H, NCH and NH ), 2.90-2.67 (m, $4 \mathrm{H}, 2 \mathrm{CH}_{2}$ ), 1.93-0.97 (m, 14 H , $7 \mathrm{CH}_{2}$ ) |
| 7d | $\begin{aligned} & 3433,1685 \\ & 1543,1209 \end{aligned}$ | $\begin{aligned} & 353\left(\mathrm{M}^{+}, 98\right), 296(96), \\ & 269(96), 179(17), 77 \\ & (58) \end{aligned}$ | 7.59-7.24 (m, 5H, Ar-H), 3.91 (s, 1H, NH), 2.90-2.68 (m, 4H, $2 \mathrm{CH}_{2}$ ), 1.86-1.77 (m, 4H, $2 \mathrm{CH}_{2}$ ), $1.34(\mathrm{~s}, 9 \mathrm{H}$, $3 \mathrm{CH}_{3}$ ) |
| 7e | $\begin{aligned} & 3441,1681, \\ & 1551,1252 \end{aligned}$ | $\begin{aligned} & 375(35), 373\left(\mathrm{M}^{+}, 100\right), \\ & 331(64), 303(67), 220 \\ & (61), 179(69), 111(57) \end{aligned}$ | 7.57-7.22 (m, 4H, Ar-H), 4.00 (s, 1H, NH), 3.38-3.30 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{NCH}_{2}$ ), 2.88-2.67 (m, $4 \mathrm{H}, 2 \mathrm{CH}_{2}$ ), 1.87-1.77 (m, $\left.4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 1.54-0.83\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$ |
| 7f | $\begin{aligned} & 3447,1676 \\ & 1552,1186 \end{aligned}$ | $\begin{aligned} & 389(33), 387\left(\mathrm{M}^{+}, 88\right), \\ & 331(58), 303(70), 220 \\ & (46), 111(95), 41(100) \end{aligned}$ | $\begin{aligned} & 7.57-7.21(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 3.94(\mathrm{t}, 1 \mathrm{H}, \mathrm{~J}=4.8 \mathrm{~Hz}, \mathrm{NH}) \text {, } \\ & 3.40-3.33\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.88-2.67\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), \\ & 1.86-1.77\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), \quad 1.48-0.87(\mathrm{~m}, \quad 7 \mathrm{H}, \\ & \left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right) \end{aligned}$ |
| 7g | $\begin{aligned} & 3436,1683, \\ & 1544,1176 \end{aligned}$ | $\begin{aligned} & 375(8), 373\left(\mathrm{M}^{+}, 23\right), \\ & 331(8), 303(20), 221 \\ & (11), 179(25), 43(100) \end{aligned}$ | 7.57-7.20 (m, 4H, Ar-H), 4.24-4.16 (m, 1H, NCH), $3.76(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{NH}), 2.88-2.67\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right)$, $1.87-1.76\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 1.12\left(\mathrm{~d}, 6 \mathrm{H}, \mathrm{J}=6.4 \mathrm{~Hz}, 2 \mathrm{CH}_{3}\right)$ |
| 7h | $\begin{aligned} & 3424,1685 \\ & 1541,1298 \end{aligned}$ | $\begin{aligned} & 415(9), 413\left(\mathrm{M}^{+}, 25\right), \\ & 331(54), 303(64), 179 \\ & (27), 55(100) \end{aligned}$ | 7.57-7.21 (m, 4H, Ar-H), 4.00-3.91 (m, 1H, NCH), $3.86(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{NH}), 2.88-2.67\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right)$, 1.95-0.98 (m, 14H, $7 \mathrm{CH}_{2}$ ) |
| $7 \mathbf{i}$ | $\begin{aligned} & 3445,1680, \\ & 1553,1211 \end{aligned}$ | $\begin{aligned} & 389(39), 387\left(\mathrm{M}^{+}, 100\right), \\ & 331(94), 303(59), 111 \\ & (13) \end{aligned}$ | 7.56-7.19 (m, 4H, Ar-H), 3.85 (s, 1H, NH), 2.88-2.67 (m, 4H, $2 \mathrm{CH}_{2}$ ), 1.85-1.77 (m, 4H, $2 \mathrm{CH}_{2}$ ), $1.35(\mathrm{~s}, 9 \mathrm{H}$, $3 \mathrm{CH}_{3}$ ) |
| $7 \mathbf{j}$ | $\begin{aligned} & 3431,1683 \\ & 1531,1172 \end{aligned}$ | $\begin{aligned} & 353\left(\mathrm{M}^{+}, 35\right), 311(14), \\ & 283(26), 221(12), 133 \\ & (36), 91(100) \end{aligned}$ | $\begin{aligned} & \text { 7.38-7.12 (m, 4H, Ar-H), 4.24-4.17 (m, 1H, NCH), } \\ & 3.88(\mathrm{~d}, 1 \mathrm{H}, \mathrm{~J}=7.2 \mathrm{~Hz}, \mathrm{NH}), 2.89-2.68\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), \\ & 2.44\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{3}\right), 1.85-1.77\left(\mathrm{~m}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 1.11(\mathrm{~d}, \\ & \left.6 \mathrm{H}, \mathrm{~J}=6.4 \mathrm{~Hz}, 2 \mathrm{CH}_{3}\right) \end{aligned}$ |

The above synthetic method provides a selective synthesis of 2-alkylamino substituted 5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones via $\mathrm{EtO}^{-} \mathrm{Na}^{+}$catalyzed reaction of functionalized carbodiimides with various aliphatic primary amines. Due to the easily accessible and versatile starting material, this method has the potential in synthesis of many biologically and pharmaceutically active thienopyrimidinones derivatives.

General Procedure for selective synthesis of 2-alkylamino-5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones 7: To a solution of iminophosphorane 4 (2.42 $\mathrm{g}, 5 \mathrm{mmol})$ in dry methylene dichloride ( 15 mL ) was added aromatic isocyanate ( 5 mmol ) under nitrogen at $0-5^{\circ} \mathrm{C}$. After the reaction mixture was stood for 12 hours at $0-5^{\circ} \mathrm{C}$, the solvent was removed off under reduced pressure and ether/petroleum ether ( $1: 2,30 \mathrm{~mL}$ ) was added to precipitate triphenylphosphine oxide, which was filtered, the solvent was removed to give carbodiimide 5,5 can be used directly without further purification. To the solution of 5 in methylene dichloride ( 15 mL ) was added alkylamine ( 5 mmol ). After the reaction mixture was stood for 10 minutes, the solvent was removed and anhydrous ethanol ( 10 mL ) with several drops of EtONa in EtOH was added ( $\mathrm{pH}=10$ ). The mixture was stirred for $6-12 \mathrm{hr}$ at room temperature. The solution was condensed and the residual was recrystallized from ethanol to give 2 -alkylamino-5, 6, 7, 8-tetrahydrobenzothieno[2, 3-d]pyrimidin-4(3H)-ones 7. The yields of 7 based on iminophosphorane 4 are listed in Table 1.

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